

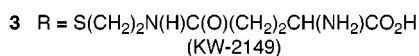
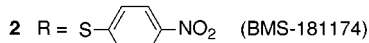
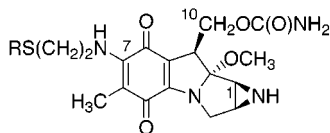
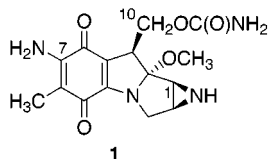
Studies on the Mode of Action of Mitomycin C(7) Aminoethylene Disulfides (BMS-181174 and KW-2149): Reactivity of 7-*N*-(Mercaptoethyl)mitomycin C

Shuang Wang and Harold Kohn*

Department of Chemistry, University of Houston,
Houston, Texas 77204-5641

Received December 4, 1998

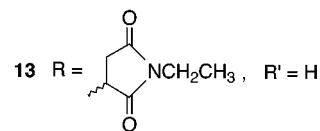
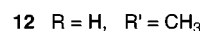
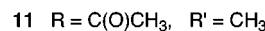
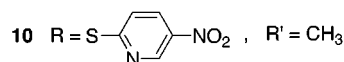
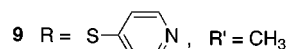
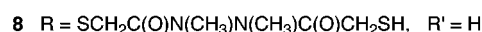
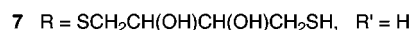
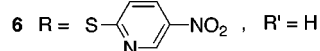
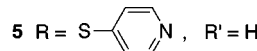
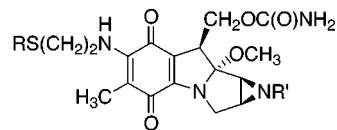
Mitomycin C (**1**) is extensively used in combination therapy to treat various neoplasms.¹ Its associated toxicities have led to an active drug development program² and the subsequent discovery of the two C(7) aminoethylene disulfides BMS-181174³ (**2**) and KW-2149⁴ (**3**). Both **2**^{5a,b} and **3**^{5c} exhibit improved pharmacological activity, compared with **1**, and **3** is currently undergoing clinical trials.^{5c}



Compounds **2** and **3** only differ from mitomycin C (**1**) in the C(7) substituent. In **2** and **3**, a substituted aminoethylene disulfide unit replaces the C(7) amino group found in **1**. Novel mechanisms^{6–8} proposed for **2** and **3** differ from the bioreductive activation pathway commonly accepted for mitomycin C.⁹ A major contention of these hypotheses is that the C(7) aminoethylene disulfide unit in **2** and **3** undergoes thiol-mediated (e.g., R'SH = cysteine, glutathione (GSH)) disulfide exchange to give **4** and R'SSR.^{6,7} Thiol **4** has never been identified. We present here preparative routes to **4** and related compounds and report on the reactivity of these species. The properties observed for thiol **4** require us to question these hypotheses^{6,7} for **2** and **3**, and they lead us to suggest another pathway.

Our approach to **4** was to prepare mitomycins that rapidly and efficiently convert to **4** under mild conditions. Two strategies were used. The first entailed synthesizing C(7)-substituted mitomycins that undergo selective disulfide cleavage. We prepared **5**^{10a,11} and **6**.^{10b} Each contained a pyridyl disulfide group, which upon treatment with either *D,L*-dithiothreitol¹² (DTT) or *N,N*-dimethyl-*N,N*-bis(mercaptoacetyl)hydrazine¹³ (DMH) underwent selective disulfide cleavage to give **7** and **8**, respectively. Subsequent intramolecular disulfide cleav-

age provided **4** and the oxidized cyclic disulfide. We expanded our study to include the two porfiromycin (R' = CH₃) analogues **9** and **10**. In the second approach, we incorporated a C(7) terminal thiol ester unit to give **11**.¹⁴ This route takes advantage of the relative ease with which thiol esters undergo base-mediated cleavage¹⁵ and yielded the porfiromycin thiol **12**.



Thiol formation was monitored by HPLC (200–400 nm, photodiode array detection).¹⁶ In Figure 1A we provide the HPLC profile (365 nm) obtained from a deaerated (Ar) methanolic (–78 °C) solution containing **6** and DTT (10 equiv). We observed the complete consumption of **6** (32.1 min) and the appearance of multiple peaks (3–4) between 22 and 26 min.¹⁷ The same peaks were observed when we used DMH in place of DTT, acetone for methanol, and **5** in place of **6**. Figure 1B provides the corresponding HPLC chromatogram for a deaerated (Ar) methanolic NaOMe solution (–78 °C) containing **11**. Significantly, multiple peaks were observed between 23 and 29 min. The same peaks were observed for DTT-treated methanolic solutions containing either **9** or **10**.¹⁷ We attributed the increased retention times for the porfiromycin multiple peaks, compared with the corresponding mitomycin peaks, to the effect of the N(1a) methyl group on the elution times.¹⁸ These experiments indicated that a comparable set of intermediates is observed in the HPLC chromatograms independent of the activation procedure (DTT (DMH), NaOMe), the solvent used (methanol, acetone), and the structure of the starting mitomycins (**5**, **6**, **9**–**11**).

Information concerning the identity of the multiple HPLC peaks was gathered through thiol trapping-experiments (4,4'-dipyridyl disulfide (DPDS), 2,2'-dithiobis(5-nitropyridine) (DTNP), *N*-ethylmaleimide

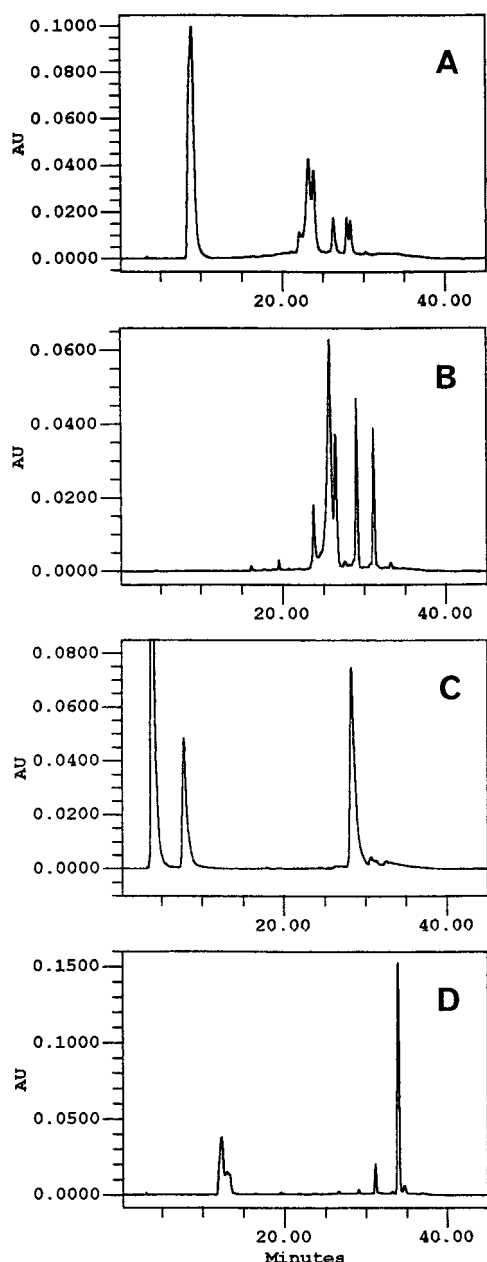
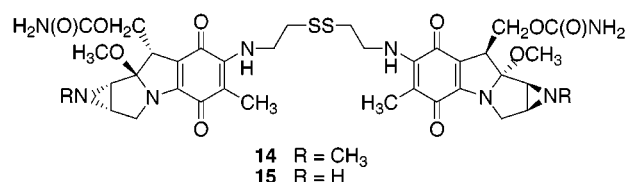


Figure 1. HPLC (365 nm) of mitomycin **6** and porfiromycin **11** activated reactions. (A) Mitomycin **6** treated with DTT (10 equiv) in MeOH at -78°C . Major peaks (min): t_{R} 8.3, 5-nitro-2-thiopyridone; 21.9, 23.0, 23.6, **4**; 26.6, **7**; 28.4, 2:1 **6**:DTT adduct; 28.8, **15**. (B) Porfiromycin **11** treated with NaOMe in MeOH at -78°C . Major peaks (min): t_{R} 23.9, 25.9, 26.6, 29.1, **12**; 31.2, **14**. (C) Mitomycin **6** sequentially treated with DTT (10 equiv) in MeOH at -78°C (Figure 1A) followed by DPDS. Major peaks (min): t_{R} 3.8, 4-thiopyridone; 8.3, 5-nitro-2-thiopyridone; 28.5, **5**. (D) Porfiromycin **11** sequentially treated with NaOMe in MeOH at -78°C (Figure 1C) followed by DTNP. Major peaks (min): t_{R} 12.3, 5-nitro-2-thiopyridone byproduct(s); 31.2, **14**; 33.6, **10**. The identities of **5** and **10** in Figure 1C,D, respectively, were confirmed by co-injection (cospot) of an authentic sample with the reaction solution in the HPLC (TLC).

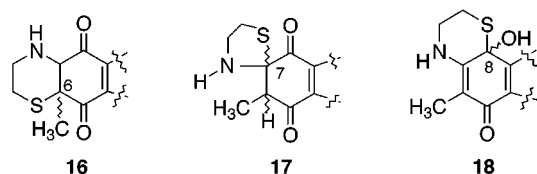
(NEM). Figure 1C shows that adding DPDS to a DTT-treated solution of **6** (Figure 1A) completely eliminated the multiple HPLC peaks and the production of **5** (28.5 min). Correspondingly, treating the methanolic NaOMe solution-containing **11** (Figure 1B) with DTNP gave **10** (33.6 min) as the major product (Figure 1D). Similarly, we found that **5** was converted to **6**, **9** to **10**, and **10** to

9 upon successive treatment with DTT and the appropriate disulfide (DPDS, DTNP) and that **11** was converted to **9** upon sequential treatment with methanolic NaOMe and DPDS. Replacing disulfides DPDS and DTNP with NEM in the thiol-trapping reactions beginning with either **5** or **6** gave diastereomeric **13**. These collective experiments demonstrated that the multiple HPLC peaks are likely to be free thiol **4** (**12**) and isomeric forms of **4** (**12**).¹⁹

Additional experiments supported this notion. Elevating the reaction temperature of a deaerated methanolic NaOMe solution of **11** from -78°C to room temperature led to near quantitative production of disulfide **14**.^{20a} Correspondingly, treatment of deaerated methanolic ("pH" 5.5, 6.5, 7.4) solutions containing either **5** or **6** at room temperature with DTT (1 equiv) gave disulfide **15**^{20a,b} as the major product. Absent in these reactions was the production of noticeable amounts of aziridine ring-opened mitosenes (HPLC analysis).



From where did the multiple peaks in the HPLC chromatograms (Figures 1A,B) come? Without NMR structural evidence we attributed the multiple peaks, in part, to **4** (**12**) and isomeric forms of the free thiol **4** (**12**), which can include C(6)-**16**,⁶ C(7)-**17**, and C(8)-**18**⁷ cyclized adducts.



A significant finding of this study was the observation of multiple peaks in the HPLC chromatograms corresponding to thiol **4** (**12**). Furthermore, we found that **4** (**12**) generation did *not* lead to aziridine ring-opened mitosenes. This finding was surprising since it has been proposed that the terminal thiol unit in **4** (**12**) initiates conversion of the mitomycin ring system to an activated mitosene and subsequent nucleophilic (DNA) attack at the C(1) and C(10) sites.^{6,7} Last, we found that **4** and **12** were efficiently converted to dimeric mitomycins **15** and **14**, respectively, at room temperature. The efficiency of **14** and **15** production may be clinically significant. Compound **15** is the major metabolite produced upon administration of **3** to normal and tumor-bearing mice.²¹ Several mechanisms exist for the anaerobic dimerization of **4** to either the bis-semiquinone or mixed hydroquinone–quinone species corresponding to **15**. Formation of *reduced* **15** from either **2** or **3** leads to a novel DNA cross-linking agent in which DNA modification of complementary strands can initially proceed at the C(1) aziridine sites on the *two* mitomycin subunits. By comparison, mitomycin C cross-linking transformations require activation of the C(1) aziridine site and the C(10) position. Chemical studies have

documented the diminished reactivity of the C(10) site in reductively activated **1** versus the C(1) position.²² Accordingly, activated dimeric mitomycins, such as reduced **15** (**14**), may permit DNA cross-linking reactions to proceed more efficiently than conventional monomeric mitomycins. Studies on the mechanistic details of this projected pathway and the sequence selectivity and structure of dimeric mitomycin–DNA adduct(s) are in progress.

Acknowledgment. This work was funded by NIH Grant RO1CA29756 and the Robert A. Welch Foundation (E-607). We gratefully acknowledge Dr. M. Kasai and Kyowa Hakko Kogyo Co., Ltd. for the generous gift of mitomycin A.

References

- (1) Carter, S. K.; Crooke, S. T. *Mitomycin C. Current Status and New Developments*; Academic Press: New York, 1979.
- (2) (a) Remers, W. A.; Dorr, R. T. In *Alkaloids: Chemical and Biological Perspective*; Pelletter, S. W., Ed.; Wiley and Sons: New York, 1988; Vol. 6, pp 1–74. (b) Remers, W. A. *The Chemistry of Antitumor Antibiotics*; Wiley and Sons: New York, 1979; Vol. 1, pp 221–276. (c) Bradner, W. T.; Remers, W. A.; Vyas, D. M. Structure–Activity Comparison of Mitomycin C and Mitomycin A Analogues (Review). *Anticancer Res.* **1989**, *9*, 1095–1100.
- (3) Vyas, D. M.; Chiang, Y.; Benigni, D.; Rose, W. C.; Bradner, W. T.; Doyle, T. W. In *Recent Advances in Chemotherapy. Anticancer Section 1*; Ishigami, J., Ed.; University of Tokyo Press: Tokyo, 1985; pp 485–486.
- (4) Kono, M.; Saitoh, Y.; Kasai, M.; Sato, A.; Shirahata, K.; Morimoto, M.; Ashizawa, T. Synthesis and Antitumor Activity of a Novel Water Soluble Mitomycin Analogue; 7-*N*-[2-[[2-(γ -L-Glutamylamino)ethyl]dithio]ethyl]mitomycin C. *Chem. Pharm. Bull.* **1989**, *37*, 1128–1130.
- (5) (a) Xu, B. H.; Singh, S. V. Effect of Buthionine Sulfoximine and Ethacrynic Acid on Cytotoxic Activity of Mitomycin C Analogues BMY 25282 and BMY 25067. *Cancer Res.* **1992**, *52*, 6666–6670. (b) Yen, W.-C.; Au, J. L.-S. Pharmacodynamic Evaluation of Mitomycin C Analogue BMS-181174 for Potential use in Intravesical Bladder Cancer Therapy. *Pharm. Res.* **1997**, *14*, 241–245. (c) Dirix, L.; Catimel, G.; Koier, I.; Provè, A.; Schrijvers, D.; Joossens, E.; De Bruijn, E.; Ardiet, C.; Evens, E.; Dumortier, A.; Clavel, M.; Van Oosterom, A. Phase I and Pharmacokinetic Study of a Novel Mitomycin C Analogue KW-2149. *Anti-Cancer Drugs* **1995**, *6*, 53–63.
- (6) He, Q.-Y.; Maruenda, H.; Tomasz, M. Novel Bioreductive Activation Mechanism of Mitomycin C Derivatives Bearing a Disulfide Substituent in Their Quinone. *J. Am. Chem. Soc.* **1994**, *116*, 9349–9350.
- (7) Kohn, H.; Wang, S. Studies on the Mechanism of Activation of C(7) Ethylenediamine Substituted Mitomycins. Relevance to the Proposed Mode of Action of BMY-25067 and KW-2149. *Tetrahedron Lett.* **1996**, *37*, 2337–2340.
- (8) For additional mechanistic studies of **3**, see: (a) Lee, J.-H.; Naito, M.; Tsuruo, T. Nonenzymatic Reductive Activation of 7-*N*-[2-[[2-(γ -L-Glutamyl-amino)ethyl]dithio]ethyl]mitomycin C by Thiol Molecules: a Novel Mitomycin C Derivative Effective on Mitomycin C-resistant Tumor Cells. *Cancer Res.* **1994**, *54*, 2398–2403. (b) Masters, J. R. W.; Know, R. J.; Hartley, J. A.; Kelland, L. R.; Hendriks, H. R.; Connors, T. KW-2149 (7-[2-(γ -L-Glutamyl-aminoethyl]dithioethyl]mitomycin C) a New Mitomycin C Analogue Activated by Serum. *Biochem. Pharmacol.* **1997**, *53*, 279–285. (c) McAdam, S. R.; Knox, R. J.; Hartley, J. A.; Masters, J. R. W. KW-2149 (7-*N*-[2-(γ -L-Glutamyl-amino)ethyl]dithioethyl]mitomycin C): DNA Interactions and Drug Uptake Following Serum Activation. *Biochem. Pharmacol.* **1998**, *55*, 1777–1783.
- (9) (a) Iyer, V. N.; Szybalski, W. Mitomycins and Porfomycin: Chemical Mechanism of Activation and Cross-linking of DNA. *Science* **1964**, *145*, 55–58. (b) Moore, H. W.; Czerniak, R. Naturally Occurring Quinones as Potential Bioreductive Alkylating Agents. *Med. Res. Rev.* **1981**, *1*, 249–280. (c) Schiltz, P.; Kohn, H. Studies on the Reactivity of Reductively Activated Mitomycin C. *J. Am. Chem. Soc.* **1993**, *115*, 10510–10518. (d) Kumar, G. S.; Lipman, R.; Cummings, J.; Tomasz, M. Mitomycin C-DNA Adducts Generated by DT-Diaphorase. Revised Mechanism of the Enzymatic Reductive Activation of Mitomycin C. *Biochemistry* **1997**, *36*, 14128–14136.
- (10) (a) Vyas, D. M.; Chiang, Y.; Doyle, T. W. Amino Disulfide Thiol Exchange Products. U.S. Patent 4,866,180, Sept 12, 1989; *Chem. Abstr.* **1998**, *108*, 150167x. (b) Vyas, D. M.; Chiang, Y.; Doyle, T. W. Amino Disulfides. U.S. Patent 4,803,212, Feb 7, 1989; *Chem. Abstr.* **1985**, *102*, 131829z.
- (11) Satisfactory spectral data was obtained for all previously prepared (**5**,^{10a} **6**,^{10b} **11**,¹⁴ **14**,^{20a} **15**^{20a,b}) and new (**5**, **6**, **13**) compounds. Compounds **7** and **8** were too unstable to isolate. Treatment of either **5** or **6** with GSH provided the corresponding GSH–mitomycin mixed disulfide 7-*N*-[2-(glutathionedithio)ethyl]mitomycin C. Additional details can be found in Wang, S. Ph.D. Thesis, University of Houston, Houston, TX, 1998.
- (12) Cleland, W. W. Dithiothreitol, a New Protective Reagent for SH Groups. *Biochemistry* **1964**, *3*, 480–482.
- (13) Singh, R.; Whitesides, G. M. A Reagent for Reduction of Disulfide Bonds in Proteins that Reduces Disulfide Bonds Faster Than Does Dithiothreitol. *J. Org. Chem.* **1991**, *56*, 2332–2337.
- (14) Kasai, M.; Sato, Y.; Kono, M.; Akira, S.; Hiroshi, S.; Kunikatsu, S.; Makoto, M.; Tadashi, A. Pharmacologically Active Mitomycin Derivatives. Eur. Pat. Appl. EP0197,799, Oct 15, 1986; *Chem. Abstr.* **1987**, *106*, 49881j.
- (15) (a) Patai, S. *The Chemistry of the Thiol Group*; Wiley: New York, 1974; p 677. (b) Maskill, H. *The Physical Basis of Organic Chemistry*; Oxford University Press: Oxford, 1985; p 159.
- (16) The HPLC analyses were conducted with the following Waters Associate Units: 515 pump (pump A), 515 pump (pump B), Millennium chromatography manager, Waters 996 photodiode array detector, Rehdodyne 7725i manual injector, C₁₈ μ Bondapak (stainless steel) column (3.9 \times 300 mm) using the following linear gradient 90% A (aqueous 0.025 M triethylammonium acetate, pH 6.5), 10% B (acetonitrile) isocratic for 5 min, then from 90% A, 10% B to 45% A, 55% B in 30 min. The flow rate was 1 mL/min, and the eluent was monitored from 200 to 400 nm.
- (17) The identities of the major remaining peaks were determined after PTLC isolation, spectroscopic elucidation, and then subsequent confirmation of the product peak in the reaction mixture by co-injection (cospot) of the authentic sample with the reaction solution in the HPLC (TLC).
- (18) A similar difference in the HPLC retention times was observed for mitomycin C (16.7 min) and porfomycin (18.1 min).
- (19) The HPLC data does not exclude multimeric adducts of **4** (**12**).
- (20) (a) Kono, M.; Saitoh, Y.; Kasai, M.; Shirahata, K.; Morimoto, M.; Ashizawa, T. Synthesis and Structure–Activity Relationships of New Dimeric Mitomycin Derivatives; 7-*N*,7'-*N*-Bis(ω -thioalkyl)dimitomycins. *J. Antibiot.* **1983**, *46*, 1428–1438. (b) Senter, P. D.; Langley, D. R.; Manger, W. E.; Vyas, D. Reassignment of the Structure for the Antitumor Agent RR-150. *J. Antibiot.* **1988**, *41*, 199–201.
- (21) (a) Kobayashi, S.; Ushiki, J.; Takai, K.; Okumura, S.; Kono, M.; Kasai, M.; Gomi, K.; Morimoto, M.; Ueno, H.; Hirata, T. Disposition and Metabolism of KW-2149, a Novel Anticancer Agent. *Cancer Chemother. Pharmacol.* **1993**, *32*, 143–150. (b) Ashizawa, T.; Okamoto, A.; Okabe, M.; Kobayashi, S.; Arai, H.; Saito, H.; Kasai, M.; Gomi, K. Characteristics of the Antitumor Activity of M-16 and M-18, Major Metabolites of a New Mitomycin C Derivative KW-2149, in Mice. *Anti-Cancer Drugs* **1995**, *6*, 763–770.
- (22) (a) Hornemann, U.; Keller, P. J.; Kozlowski, J. F. Formation of 1-Ethylxanthyl-2,7-diaminomitosene and 1,10-Diethylxanthyl-2,7-diaminodecarbamoylmitosene in Aqueous Solution upon Reduction-Reoxidation of Mitomycin C in the Presence of Potassium Ethylxanthate. *J. Am. Chem. Soc.* **1979**, *101*, 7121–7124. (b) Hornemann, U.; Iguchi, K.; Keller, P. J.; Vu, H. M.; Kozlowski, J. F.; Kohn, H. Reactions of Mitomycin C with Potassium Ethyl Xanthate in Neutral Aqueous Solution. *J. Org. Chem.* **1983**, *48*, 5026–5033. (c) Danishefsky, S. J.; Egbertson, M. On the Characterization of Intermediates in the Mitomycin Activation Cascade: A Practical Synthesis of an Aziridinomitosene. *J. Am. Chem. Soc.* **1986**, *108*, 4648–4650. (d) Kohn, H.; Zein, N.; Lin, X. Q.; Ding, J.-Q.; Kadish, K. M. Mechanistic Studies on the Mode of Reaction of Mitomycin C under Catalytic and Electrochemical Reductive Conditions. *J. Am. Chem. Soc.* **1987**, *109*, 1833–1840. (e) Bean, M.; Kohn, H. Studies on the Reaction of Mitomycin C with Potassium Ethyl Monothiocarbonate under Reductive Conditions. *J. Org. Chem.* **1983**, *48*, 5033–5041. (f) Siegel, D.; Gibson, N. W.; Preusch, P. C.; Ross, D. Metabolism of Mitomycin C by DT-Diaphorase: Role in Mitomycin C-induced DNA Damage and Cytotoxicity in Human Colon Carcinoma Cells. *Cancer Res.* **1990**, *50*, 7843–7849.